

Extraction of Tannins from Yellow Birch: Enhanced Process for Water Conservation and Energy Savings

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Supporting information

1. Water and energy consumption for tannin extraction from 1 kg of dried bark using the hot water extraction method

1.1 Water consumption

Moisture of bark = 30%

Mass of bark with moisture used per batch = 0.80 kg corresponding to 0.56 kg of dried bark and 0.24 kg of water

Water / dried bark ratio = 2.5, therefore the mass of water used for the extraction = $(2.5 * 0.56 \text{ kg}) + 0.24 \text{ kg} = 1.64 \text{ kg}$

Mass of water required for 1 kg of dried bark = $1.64 / 0.56 = 2.93 \text{ kg}$

For 0.56 kg of dried bark, 1.2 kg of hot water was required for the residue washing

For 1 kg of dried bark, the mass of water required for the residue washing = $1.2 / 0.56 \sim 2.15 \text{ kg}$

The water consumption for 1 kg of dried bark = $2.93 + 2.15 = \mathbf{5.08 \text{ kg}}$

1.2 Energy consumption

1.2.1 Chopping step

A chopper machine Pierret G28L1 with a power of 4 kW was used for the chopping step.

145.7 kg of dried bark were chopped in 1.33 hour

Energy consumption for the chopping step = $4 * 1.33 / 145.7 \sim \mathbf{0.04 \text{ kWh/kg}}$

1.2.2 Grinding step

A granulator Mo-Di-Tec GPlus with the following characteristics was used for the grinding step.

- Power = 1.5 kW
- $\text{Cos}(\Phi) = 0.87$

Energy consumption for the grinding step, $E_g = \text{Power} * \text{cos}(\Phi) * \text{hour} / \text{mass of bark}$

Grinding iteration	Mass of dried bark (kg)	Grinding time (min)	E_g (kWh/kg)
1 time	3.25	21.50	0.15
2 times	3.30	13.60	0.24
3 times	3.35	12.40	0.32

1.2.3 Extraction step

- Mass heat capacity of water = $4.185 \text{ kJ. kg}^{-1}. \text{K}^{-1}$
- Mass heat capacity of bark = $1.500 \text{ kJ. kg}^{-1}. \text{K}^{-1}$
- The extraction temperature was fixed at 95°C
- The extraction time was kept at 2 hours
- $1 \text{ kJ} \sim 0.000278 \text{ kWh}$

- **Energy required for the heating**

The water and the bark were heated from 20°C to 95°C and kept at this temperature for X hours.

Energy required to heat 2.93 kg of water from 20°C to 95°C
 $= 2.93 * 4.185 * (95-20) * 0.000278 \sim 0.26 \text{ kWh}$

Energy required to heat 1 kg of dried bark from 20°C to 95°C
 $= 1 * 1.500 * (95-20) * 0.000278 \sim 0.03 \text{ kWh}$

Energy required to heat water and bark for the extraction step using 1 kg of dried bark
 $= 0.26 + 0.03 = \mathbf{0.29 \text{ kWh}}$

After extraction, the tannin solution was separated from the bark residue by centrifugation and the residue was washed 2 times using hot water.

For 1 kg of dried bark, the mass of water required for the washing step = 2.15 kg

Energy required to heat water from 20°C to 95°C
 $= 2.15 * 4.185 * (95-20) * 0.000278 \sim 0.19 \text{ kWh}$

Total energy required to heat water and bark using 1 kg of dried bark
 $= 0.29 + 0.19 = \mathbf{0.48 \text{ kWh}}$

- **Energy lost**

A reactor was used during the extraction. It was assumed that, for each hour of extraction, 5% of the energy was lost. Energy lost during 2 hours extraction, $E_l = 0.29 * 2 * 5\% \sim \mathbf{0.03 \text{ kWh}}$

The energy lost during the water heating for the residue washing was negligible.

- **Energy consumed by the mechanical stirrer**

A mechanical stirrer with a power of 87 W was used for the mixing during the extraction.
 Energy consumption for the stirrer during 2 hours using 1 kg of dried bark
 $= 0.087 * 2 \sim \mathbf{0.18 \text{ kWh}}$

- **Energy consumed during the centrifugation**

A centrifuge with a power of 1.4 kW was used to separate the residue from the tannin solution.
 For 1 kg of dried bark, the centrifuge was run during 10 min for the separation therefore the energy
 consumed $= 1.4 * 10 / 60 \sim \mathbf{0.24 \text{ kWh}}$

Total energy consumption for the extraction step using 1 kg of dried bark
 $= 0.48 + 0.03 + 0.18 + 0.24 = \mathbf{0.93 \text{ kWh}}$

1.2.4 Evaporation step

Latent heat of vaporization of water = 2257 kJ/kg

The total mass of water used for the extraction = 5.08 kg

After tannin extraction and separation, 15% of the total mass of water was lost therefore the mass
 of water to be evaporated to recover the tannin powder $= 5.08 - 5.08 * 15\% \sim 4.32 \text{ kg}$.

The temperature of the tannin solution after separation was 70°C.

To evaporate the water, it is heated until 100°C and transformed into vapor.

Energy required to evaporate the water and to recover the tannin powder. 34.63

$= (4.32 * 4.185 * (100 - 70)) + (4.32 * 2257) = 10292.616 \text{ kJ}$

$= 10292.616 * 0.000278 \sim \mathbf{2.86 \text{ kWh}}$

Total energy consumption for tannin extraction from 1 kg of dried bark using the hot water
 extraction method $= 0.04 + 0.15 + 0.93 + 2.86 = \mathbf{3.98 \text{ kWh}}$

2. Water and energy consumption for the production of 1 kg of tannin

2.1 Hot water extraction method

- **Water consumption for an extraction time of 2 h**

Tannin yield = 2.30 %

Mass of dried bark required for the production of 1 kg of tannin $= 100 / 2.30 = 43.48 \text{ kg}$

Mass of required for the production of 1 kg of tannin = mass of dried bark required for the
 production of 1 kg of tannin *multiplied* by the mass of water used for the extraction of 1 kg of dried
 bark $= 43.48 * 5.08 \sim \mathbf{220.88 \text{ kg}}$

- **Energy consumption for an extraction time of 2 h**

Energy consumption for the chopping step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the chopping energy for 1 kg of dried bark = $43.48 * 0.04 \sim 1.74 \text{ kWh}$

Energy consumption for the grinding step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the grinding energy for 1 kg of dried bark = $43.48 * 0.15 \sim 6.52 \text{ kWh}$

Energy consumption for the extraction step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the extraction energy for 1 kg of dried bark = $43.48 * 0.93 \sim 40.44 \text{ kWh}$

Energy consumption for the isolation step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the isolation energy for 1 kg of dried bark = $43.48 * 2.86 \sim 124.35 \text{ kWh}$

Total energy consumption for the production of 1 kg tannin using the hot water extraction method = $1.74 + 6.52 + 40.44 + 124.35 = 173.05 \text{ kWh}$

2.2 Method 1: Combination of the hot water extraction and the ultrasound assisted extraction

- **Water consumption for an extraction time of 1 h**

Tannin yield = **3.20 %**

Mass of the dried bark required for the production of 1 kg of tannin = $100 / 3.20 \sim 31.25 \text{ kg}$

Water consumption for the production of 1 kg of tannin = mass of dried bark required for the production of 1 kg of tannin *multiplied* by the mass of water used for the extraction of 1 kg of dried bark = $31.25 * 5.08 \sim 158.75 \text{ kg}$

- **Energy consumption for an extraction time of 1 h**

Energy consumption for the chopping step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the chopping energy for 1 kg of dried bark = $31.25 * 0.04 \sim 1.25 \text{ kWh}$

Energy consumption for the grinding step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the grinding energy for 1 kg of dried bark = $31.25 * 0.15 \sim 4.69 \text{ kWh}$

Energy consumption for the extraction step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the extraction energy for 1 kg of dried bark using Method 1

Calculation of the extraction energy for 1 kg of dried bark using Method 1

The extraction time of Method 1 is divided into two parts:

- Extraction time of 0.75 hour using the hot water extraction method
- Extraction time of 0.25 hour using an ultrasound probe with a power of 0.04 kW

Energy required for the heating (water and bark) is the same as that of the hot water method = 0.48 kWh

Energy consumed by the mechanical stirrer = $0.087 * 0.75 \sim 0.07$ kWh

Energy lost = $0.29 * 0.75 * 5\% \sim 0.01$ kWh

Energy consumed by the ultrasound probe = $0.04 * 0.25 / 0.56 \sim 0.02$ kWh

Energy consumed by the centrifuge = 0.24 kWh

Energy consumption for the extraction step for 1 kg of dried bark using Method 1
= $0.48 + 0.07 + 0.01 + 0.02 + 0.24 = 0.82$ kWh

Energy consumption for the extraction step for 31.3 kg of dried bark using Method 1
= $31.25 * 0.82 \sim \mathbf{25.63}$ kWh

Energy required for the isolation step = $2.86 * 31.25 \sim \mathbf{89.38}$ kWh

Total energy consumption for the production of 1 kg of tannin using Method 1
= $1.25 + 4.69 + 25.63 + 89.38 = \mathbf{120.95}$ kWh

- **Water consumption for an extraction time of 2 h**

Tannin yield = **3.50 %**

Mass of the dried bark required for the production of 1 kg of tannin = $100 / 3.50 \sim \mathbf{28.57}$ kg

Mass of water used for the production of 1 kg of tannin = mass of dried bark required for the production of 1 kg of tannin *multiplied* by the mass of water used for the extraction of 1 kg of dried bark = $28.57 * 5.08 \sim \mathbf{145.14}$ kg

- **Energy consumption for an extraction time of 2 h**

Energy consumption for the chopping step = Mass of dried bark required to produce 1 kg of tannin *multiplied* by the chopping energy for 1 kg of dried bark = $28.57 * 0.04 \sim \mathbf{1.14}$ kWh

Energy consumption for the grinding step = Mass of dried bark required to produce 1 kg of tannin *multiplied* by the grinding energy for 1 kg of dried bark = $28.57 * 0.15 \sim \mathbf{4.29}$ kWh

Energy consumption for the extraction step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the extraction energy for 1 kg of dried bark using Method 1

The extraction time of the method 1 is divided into two parts:

- Extraction time of 1.75 hour using the hot water method
- Extraction time of 0.25 hour using a ultrasound probe with a power of 0.04 kW

Energy required for the heating is the same as the hot water method = 0.48 kWh

Energy consumed by the mechanical stirrer = $0.087 * 1.75 \sim 0.16$ kWh

Energy lost = $0.29 * 1.75 * 5\% \sim 0.03$ kWh

Energy consumed by the ultrasound probe = $0.04 * 0.25 / 0.56 \sim 0.02$ kWh

Energy consumed by the centrifuge = 0.24 kWh

Energy consumption for the extraction step for 1 kg of dried bark using Method 1
= $0.48 + 0.16 + 0.03 + 0.02 + 0.24 = 0.93$ kWh

Energy consumption for the extraction step using 28.57 kg of dried bark = $28.57 * 0.93 \sim 26.57$ kWh

The energy required for the isolation step = $2.86 * 28.47 = 81.71$ kWh

Total energy consumption for the production of 1 kg of tannin using Method 1
= $1.14 + 4.29 + 26.57 + 81.71 = 113.71$ kWh

2.3 Method 2 : Extract solution of the first extraction is recycled 1 time using the hot water method

2.3.1 Water consumption for an extraction time of 2 h

Tannin yield = **2.20 %**

Mass of the dried bark required for the production of 1 kg of tannin = $100 / 2.20 \sim 45.45$ kg

Mass of water required for the first extraction = $45.45 * 5.08 \sim 230.89$ kg

After the separation of the tannin solution from the bark residue, 15% of the initial mass of the water was not recovered. Therefore a mass of hot water equivalent to $230.89 * 15\% \sim 34.63$ kg was added to the collected tannin solution after separation and used as solvent in a second extraction.

The hot water used to wash the residue of the first extraction was also used to wash the residue of the second extraction therefore the mass of water required for the second extraction = 0.

Total mass of water used for the production of 1 kg of tannin when the extract solution is reused 1 time = $(230.89 + 34.63) / 2 \sim 132.76$ kg

Estimation E1: Extract solution of the first extraction recycled 2 times using the hot water method

It was assumed that the tannin yield obtained when the extract solution is reused 2 times is the same as when the extract solution is reused 1 time, therefore:

Tannin yield = 2.20 %

Mass of the dried bark required for the production of 1 kg of tannin = 45.45 kg

Mass of water used for the first extraction = 230.89 kg

Additional mass of water used for the second extraction = 34.63 kg

Additional mass of water used for the third extraction is the same as that of the second extraction = 34.63 kg

The hot water used to wash the residue of the first and second extraction was also used to wash the residue of the third extraction therefore the mass of water required for the residue washing of the third extraction = 0.

Total mass of water used for the production of 1 kg of tannin when the extract solution is reused 2 times = $(230.89 + 34.63 + 34.63) / 3 = 100.05$ kg

Estimation E2: Extract solution of the first extraction recycled 3 times using the hot water method

It was assumed that the tannin yield obtained when the extract solution is reused 3 times is the same as when the extract solution is reused 1 time, therefore:

Tannin yield = 2.20 %

Mass of water used for the first extraction = 230.89 kg

Additional mass of water used for the second extraction = 34.63 kg

Additional mass of water used for the third extraction = 34.63 kg

Additional mass of water used for the fourth extraction is the same as that of the third extraction = 34.63 kg

The hot water used to wash the residue of the first, second and third extraction was also used to wash the residue of the fourth extraction therefore the mass of water required for the residue washing of the fourth extraction = 0.

Total mass of water used for the production of 1 kg of tannin when the extract solution is reused 3 times = $(230.89 + 34.63 + 34.63 + 34.63) / 4 \sim \mathbf{83.70 \text{ kg}}$

2.3.2 Energy consumption for an extraction time of 2 h

Energy consumption for the chopping step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the chopping energy for 1 kg of dried bark = $45.45 * 0.04 \sim \mathbf{1.82 \text{ kWh}}$

Energy consumption for the grinding step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the grinding energy for 1 kg of dried bark = $45.45 * 0.15 \sim \mathbf{6.82 \text{ kWh}}$

Energy consumption for the extraction step = (Energy consumption for the first extraction + Energy consumption for the second extraction) / 2

Energy consumption of the first extraction = $0.93 * 45.45 = 42.27 \text{ kWh}$

Calculation of the energy consumption for the second extraction

34.63 kg of hot water were added during the second extraction

Energy required to heat 34.63 kg of water = $34.63 * 4.185 * (95 - 20) * 0.000278 \sim 3.02 \text{ kWh}$

Energy required to heat 1 kg of dried bark = 0.03 kWh

Energy required to heat 45.45 kg of bark = $45.45 * 0.03 \sim 1.36 \text{ kWh}$

The recovered tannin solution of the first extraction needs to be heated from 70°C to 95°C because during the separation step the temperature drops.

Energy required to heat the recovered tannin solution

= $(230.89 - 230.89 * 15\%) * 4.185 * (95 - 70) * 0.000278 \sim 5.71 \text{ kWh}$

The hot water used to wash the residue of the first extraction was also used to wash the residue of the second extraction therefore the water heating energy for the residue washing for the second extraction = 0.

Energy consumption during the second extraction step (losses, mechanical stirrer and centrifuge) = $45.45 * (0.03 + 0.18 + 0.24) \sim 20.45 \text{ kWh}$

Total energy consumption of the second extraction = $3.02 + 1.36 + 5.71 + 20.45 \sim 30.54$ kWh

Energy consumption for the extraction step when the extract solution is reused 1 time
= $(42.27 + 30.54) / 2 \sim \mathbf{36.41}$ kWh

Energy required to evaporate 34.63 kg of water
= $(34.63 * 4.185 * (100 - 70)) + (34.63 * 2257) = 82507.7065$ kJ
= 82507.7065 kJ * $0.000278 \sim 22.94$ kWh

Energy required for the isolation step when the extract solution is reused 1 time
= $(2.86 * 45.45 + 22.94) / 2 \sim \mathbf{76.46}$ kWh

Total energy consumption for the production of 1 kg of tannin when the extract solution is reused 1 times = $1.82 + 6.82 + 36.41 + 76.46 = \mathbf{121.51}$ kWh

Estimation E1: Extract solution of the first extraction recycled 2 times using the hot water method

Energy consumption for the extraction step = (Energy consumption for the first extraction step + Energy consumption for the second extraction step + Energy consumption for the third extraction step) / 3

Energy consumption of the first extraction = 42.27 kWh

Energy consumption for the second extraction = 30.54 kWh

Energy consumption for the third extraction is the same as that of the second extraction = 30.54 kWh

Energy consumption for the extraction step = $(42.27 + 30.54 + 30.54) / 3 \sim \mathbf{34.45}$ kWh

Energy required to evaporate 34.63 kg of water
= $(34.63 * 4.185 * (100 - 70)) + (34.63 * 2257) = 82507.7065$ kJ
= 82507.7065 kJ * $0.000278 \sim 22.94$ kWh

Energy required for the isolation step = $(2.86 * 45.45 + 22.94 + 22.94) / 3 \sim \mathbf{58.62}$ kWh

Total energy consumption for the production of 1 kg of tannin when the extract solution is reused 2 times = $1.82 + 6.82 + 34.45 + 58.62 = \mathbf{101.71}$ kWh

Estimation E2: Extract solution of the first extraction recycled 3 times using the hot water method

Energy consumption for the extraction step = (Energy consumption for the first extraction step + Energy consumption for the second extraction step + Energy consumption for the third extraction step + Energy consumption for the fourth extraction step) / 4

Energy consumption of the first extraction step = 42.27 kWh

Energy consumption for the second extraction = 30.54 kWh

Energy consumption for the third extraction = 30.54 kWh

Energy consumption for the fourth extraction is the same as that of the third extraction = 30.54 kWh

Energy consumption for the extraction step = $(42.27 + 30.54 + 30.54 + 30.54) / 4 \sim 33.47$ kWh

Energy required to evaporate 34.63 kg of water
= $(34.63 * 4.185 * (100 - 70)) + (34.63 * 2257) = 82507.7065$ kJ
= $82507.7065 \text{ kJ} * 0.000278 \sim 22.94$ kWh

Energy required for the isolation step = $(2.86 * 45.45 + 22.94 + 22.94 + 22.94) / 4 \sim 49.70$ kWh

Total energy consumption for the production of 1 kg of tannin when the extract solution is reused 3 times = $1.82 + 6.82 + 33.47 + 49.70 = 91.81$ kWh

2.3 Method 3 : Extract solution recycled 1 time using Method 1

2.3.1 Water consumption for an extraction time of 2 h

Tannin yield = **2.60 %**

Mass of the dried bark required for the production of 1 kg of tannin = $100 / 2.60 \sim 38.46$ kg

Mass of water used for the first extraction = $38.46 * 5.08 \sim 195.38$ kg

After the separation of the tannin solution from the bark residue, 15% of the initial mass of the water was not recovered. Therefore a mass of hot water equivalent to $195.38 * 15\% \sim 29.31$ kg was added to the collected hot tannin solution and used as solvent for a second tannin extraction. The hot water used to wash the residue of the first extraction was also used to wash the residue of the second extraction therefore the water heating energy for the residue washing of the second extraction = 0.

Mass of water used for the production of 1 kg of tannin when the extract solution is reused 1 time using Method 1 = $(195.38 + 29.31) / 2 = 112.35$ kg

Estimation E3: Extract solution of the first extraction recycled 2 times using Method 1

It has been assumed that the tannin yield obtained when the extract solution is reused 2 times is the same as when the extract solution is reused 1 time, therefore:

Tannin yield = 2.60 %

Mass of the dried bark required for the production of 1 kg of tannin = $100 / 2.60 \sim 38.46$ kg

Mass of water used for the first extraction = 195.38 kg

Additional mass of water used for the second extraction = 29.31 kg

Additional mass of water used for the third extraction is the same as that of the second extraction = 29.31 kg

The hot water used to wash the residue of the first and second extraction was also used to wash the residue of the third extraction therefore the mass of water required for the residue washing of the third extraction = 0.

Mass of water used for the production of 1 kg of tannin when the extract solution is reused 2 times = $(195.38 + 29.31 + 29.31) / 3 = 84.67$ kg

Estimation E4: Extract solution of the first extraction recycled 3 times using Method 1

It has been assumed that the tannin yield obtained when the extract solution is reused 3 times is the same as when the extract solution is reused 1 time, therefore:

Tannin yield = 2.60 %

Mass of the dried bark required for the production of 1 kg of tannin = $100 / 2.60 \sim 38.46$ kg

Mass of water used for the first extraction = 195.38 kg

Additional mass of water used for the second extraction = 29.31 kg

Additional mass of water used for the third extraction = 29.31 kg

Additional mass of water used for the fourth extraction is the same as that of the third extraction = 29.31 kg

The hot water used to wash the residue of the first, second and third extraction was also used to wash the residue of the fourth extraction therefore the mass of water required for the residue washing of the fourth extraction = 0.

Mass of water used for the production of 1 kg of tannin when the extract solution is reused 3 times = $(195.38 + 29.31 + 29.31 + 29.31) / 4 \sim 70.83$ kg

2.3.2 Energy consumption for an extraction time of 2 h

Energy consumption for the chopping step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the chopping energy for 1 kg of dried bark = $38.46 * 0.04 \sim 1.54$ kWh

Energy consumption for the grinding step = Mass of dried bark required for the production of 1 kg of tannin *multiplied* by the grinding energy for 1 kg of dried bark = $38.46 * 0.15 \sim 5.77$ kWh

Energy consumption for the extraction step = (Energy consumption for the first extraction step + Energy consumption for the second extraction step) / 2

Energy consumption of the first extraction = $0.93 * 38.46 = 35.77$ kWh

Energy consumption for the second extraction

29.31 kg of hot water were added during the second extraction

Energy required to heat 29.31 kg of water = $29.31 * 4.185 * (95 - 20) * 0.000278 \sim 2.56$ kWh

Energy required to heat 1 kg of dried bark = 0.03 kWh

Energy required to heat 38.46 kg of bark = $38.46 * 0.03 \sim 1.15$ kWh

The recovered tannin solution of the first extraction needs to be heated from 70°C to 95°C because during the separation step the temperature drops.

Energy required to heat the recovered tannin solution

= $(195.38 - 29.31) * 4.185 * (95 - 70) * 0.000278 \sim 4.83$ kWh

Energy consumption for the second extraction step (mechanical stirrer, losses, ultrasound, centrifuge) = $38.46 * (0.16 + 0.03 + 0.02 + 0.24) \sim 17.31$ kWh

Total energy consumption for the second extraction = $2.56 + 1.15 + 4.83 + 17.31 \sim 25.85$ kWh

Energy consumption for the extraction step when the extract solution is reused 1 time using Method 1 = $(35.77 + 25.85) / 2 \sim \mathbf{30.81}$ kWh

Energy required to evaporate 34.63 kg of water
= $(29.31 * 4.185 * (100 - 70)) + (29.31 * 2257) = 69832.5405$ kJ
= 69832.5405 kJ * $0.000278 \sim 19.41$ kWh

Energy required for the isolation step when the extract solution is reused 1 time using Method 1 = $(2.86 * 38.46 + 19.41) / 2 \sim \mathbf{64.70}$ kWh

Total energy consumption for the production of 1 kg of tannin when the extract solution is reused 1 time using Method 1 = $1.54 + 5.77 + 30.81 + 64.70 = \mathbf{102.82}$ kWh

Estimation E3: Extract solution of the first extraction recycled 2 times using Method 1

Energy consumption for the extraction step = (Energy consumption for the first extraction step + Energy consumption for the second extraction step + Energy consumption for the third extraction step) / 3

Energy consumption for the first extraction = $0.93 * 38.46 = 35.77$ kWh

Energy consumption for the second extraction = 25.85 kWh

Energy consumption for the third extraction is the same as that of the second extraction = 25.85 kWh

Energy consumption for the extraction step when the extract solution is reused 2 times using Method 1 = $(35.77 + 25.85 + 25.85) / 3 \sim \mathbf{29.16}$ kWh

Energy required for the isolation step using Method 2 = $(19.41 + 19.41 + 2.86 * 38.46) / 3 \sim \mathbf{49.61}$ kWh

Total energy consumption for the production of 1 kg of tannin when the extract solution is reused 1 time using Method 1 = $1.54 + 5.77 + 29.16 + 49.61 = \mathbf{86.08}$ kWh

Estimation E4: Extract solution of the first extraction recycled 3 times using Method 1

Energy consumption for the extraction step = (Energy consumption for the first extraction step + Energy consumption for the second extraction step + Energy consumption for the third extraction step + Energy consumption for the fourth extraction step) / 4

Energy consumption for the first extraction = $0.93 * 38.46 = 35.77$ kWh

Energy consumption for the second extraction = 25.85 kWh

Energy consumption for the third extraction = 25.85 kWh

Energy consumption for the fourth extraction is the same as that of the third extraction = 25.85 kWh

Energy consumption for the extraction step when the extract solution is reused 2 times using Method 1 = $(35.77 + 25.85 + 25.85 + 25.85) / 4 \sim \mathbf{28.33 \text{ kWh}}$

Energy required for the isolation step using Method 2 = $(19.41 + 19.41 + 19.41 + 2.86 * 38.46) / 4 \sim \mathbf{42.06 \text{ kWh}}$

Total energy consumption for the production of 1 kg of tannin when the extract solution is reused 1 time using Method 1 = $1.54 + 5.77 + 28.33 + 42.06 = \mathbf{77.70 \text{ kWh}}$